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10/563,168	12/29/2005	Yasuyuki Goto	20441/0202716-US0	3689
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Applicat	ion No.	Applicant(s)		
Office Action Summary		10/563,	168	GOTO ET AL.		
		Examine	er	Art Unit		
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A SH WHIC - Exter after - If NC - Failu Any r	ORTENED STATUTORY PERIOD F CHEVER IS LONGER, FROM THE Masions of time may be available under the provisions SIX (6) MONTHS from the mailing date of this complete period for reply is specified above, the maximum street or reply within the set or extended period for reply eply received by the Office later than three months and patent term adjustment. See 37 CFR 1.704(b).	IAILING DATE OF T of 37 CFR 1.136(a). In no e nunication. atutory period will apply and will, by statute, cause the ap	THIS COMMUNICATION  INVENTE, however, may a reply be the service of the service o	N. mely filed the mailing date of this of the (35 U.S.C. § 133).	·	
Status						
2a)⊠	Responsive to communication(s) file This action is <b>FINAL</b> .  Since this application is in condition closed in accordance with the practi	2b)∏ This action is for allowance excep	non-final. ot for formal matters, pro		e merits is	
Dispositi	on of Claims					
5)□ 6)⊠ 7)□ 8)□ <b>Applicati</b>	Claim(s) <u>8-27</u> is/are pending in the a 4a) Of the above claim(s) is/a Claim(s) is/are allowed. Claim(s) <u>8-27</u> is/are rejected. Claim(s) is/are objected to. Claim(s) are subject to restrict on Papers The specification is objected to by th	re withdrawn from o				
10)	The specification is objected to by the The drawing(s) filed on is/are:  Applicant may not request that any objected to ather or declaration is objected to ather the specifical including the oath or declaration is objected to ather the specifical including the spe	a) accepted or bection to the drawing(s) the correction is requ	be held in abeyance. Se ired if the drawing(s) is ob	e 37 CFR 1.85(a). ojected to. See 37 C	, ,	
Priority ι	ınder 35 U.S.C. § 119					
<ul> <li>12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).</li> <li>a) All b) Some * c) None of:</li> <li>1. Certified copies of the priority documents have been received.</li> <li>2. Certified copies of the priority documents have been received in Application No</li> <li>3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).</li> <li>* See the attached detailed Office action for a list of the certified copies not received.</li> </ul>						
2)  Notic 3) Inform	t(s) e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (F nation Disclosure Statement(s) (PTO/SB/08) r No(s)/Mail Date <u>06/03/08</u> .	PTO-948)	4) Interview Summary Paper No(s)/Mail D 5) Notice of Informal F 6) Other:	ate		

#### **DETAILED ACTION**

## Response to Amendments

The amendments filed on 10/20/08 is acknowledged.

Claims 8-27 are pending in the instant application.

# Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 8 and 9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Watanabe et al (US 5543237) further in view of Lee et al (US 20040180234)

Regarding claim 8, Watanabe teaches a charge injection type electroluminescence device (see at least Fig.1 and Abstract) for undergoing luminescence by recombination of a hole to be injected from an anode and an electron to be injected from a cathode (see electrodes 2 and 5), comprising: a luminescent layer 4 formed only of an inorganic compound (See Abstract).

Watanabe does not teach that the luminescent layer is provided between a hole transport layer and an electron transport layer, each formed of an organic compound.

In the same field of endeavor of electroluminescence devices, Lee teaches a light emitting device (see at least Fig.1), wherein an electron and hole transport layer (6, 4) are formed of an organic material ([0053]-[0085]) on either side of the luminescent

layer 5 in order to efficiently transfer holes from the anode and electrons to the cathode; to the emitting layer between the electrodes to which an electric field is applied ([0053] that also applies to electron transport layer)

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Therefore, it would have been obvious to one of ordinary skill in the art, at the time of the invention, to modify the electron and hole transport layers, as disclosed by Lee, in the electroluminescence device of Watanabe in order to efficiently transfer holes from the anode and electrons from the cathode to the emitting layer between the electrodes to which an electric field is applied.

Regarding claim 9, Watanabe teaches an electroluminescence device (Fig.1), wherein the inorganic compound is provided with a metal compound (see Abstract) which undergoes luminescence by luminescent transition by spin tolerance transition or spin inhibition transition, or undergoes luminescence by luminescent transition by inner-shell transition of a metal ion.

Initially, and with respect to the phrase "which undergoes luminescence by luminescent transition by spin tolerance transition or spin inhibition transition, or undergoes luminescence by luminescent transition by inner-shell transition of a metal ion" in claim 9, it is respectfully noted that intended use and/or other types of functional language must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, which in the above claim

is "the inorganic compound is provided with a metal compound which undergoes luminescence "that is disclosed by the prior art then it meets the claim.

Claims 10-27 are rejected under 35 U.S.C. 103(a) as being unpatentable over Watanabe et al (US 5543237) and Lee et al (US 20040180234) as applied to claims 8 and 9;further in view of Leblans (US 5632930).

Regarding Claims 10 and 11, Watanabe and Lee teaches the invention set forth above (see rejection in Claims 8 and 9 above).

The previous combination is silent regarding an electroluminescence device, wherein the inorganic compound is a combination of a luminescent metal compound with an inorganic compound capable of dissolving the metal compound therein as a solid solution.

In the same field of endeavor of phosphors, Leblans teaches an inorganic phosphor for X-ray intensifying screens having composition that comprises combination of europium (II) bromide (inorganic compound) with cesium iodide (Cesium lodide is the luminescent metal compound; see lines 37-46 of col.6, lines 16-20 of col.5 and claim 7 of Leblans) in order to stabilize the device in which the phosphor is employed against humidity (see lines 1-5 of col.3).

Therefore, it would have been obvious to one of ordinary skill in the art, at the time of the invention, to modify the inorganic luminescent layer, as disclosed by Leblans, in the electroluminescence device of Kido in order to stabilize the device in which the phosphor is employed against humidity.

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Further, Leblan implicitly teaches that the inorganic compound is a combination of a luminescent metal compound with an inorganic compound capable of dissolving the metal compound therein as a solid solution. (Lines 35-52 of col.6 of Leblan; wherein the mixture is a powder).

Note: Regarding the Leblans reference that teaches **the same phosphor** for X-ray intensifying screens but does not disclose that the phosphor composition is used for hole and an electron recombination; a composition that has already been disclosed by Leblans as a **phosphor** does not become patentable upon the discovery of a new property of the phosphor that undergoes a hole and electron recombination.

See section I of 2112 [R-3] in the MPEP that states: wherein I. **SOMETHING**WHICH IS OLD DOES NOT BECOME PATENTABLE UPON THE DIS-COVERY OF

A NEW PROPERTY "[T]he discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the discoverer."

Regarding Claims 12-27, Kido teaches the invention set forth above (see rejection in Claims 8-11 above).

Kido is silent regarding an electroluminescence device,

- 1) wherein the inorganic compound is a metal halide (claims 12-13)
- 2) wherein the luminescent metal compound and the inorganic compound capable of dissolving the metal compound therein as a solid solution are both metal halides (claims 14-15).

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3) wherein the inorganic compound is a combination of a halide of a rare earth element with a halide of an alkali metal or an alkaline earth metal (claims (claims 16 - 17).

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- 4) wherein the combination of a luminescent metal compound with an inorganic compound capable of dissolving the metal compound therein as a solid solution is a combination of a halide of a rare earth element with a halide of an alkali metal or an alkaline earth metal (claims 18 and 19).
- 5) wherein the inorganic compound is a combination of a halide of divalent europium with a halide of an alkali metal or an alkaline earth metal (claims 20-21).
- 6) wherein the combination of a luminescent metal compound with an inorganic compound capable of dissolving the metal compound therein as a solid solution is a combination of a halide of a rare earth element with a halide of an alkali metal or an alkaline earth metal (claims 22 and 23).
- 6) wherein the inorganic compound is a combination of europium (II) bromide with cesium iodide (CsI) (claims 24-25).
- 7) wherein the combination of a luminescent metal compound with an inorganic compound capable of dissolving the metal compound therein as a solid solution is a combination of europium (II) bromide with cesium iodide (claims 26 and 27).

In the same field of endeavor, Leblans teaches a phosphor composition that comprises combination of europium(II) bromide (for claims 24-27; a halide of divalent europium (rare earth element) for claims 12-23) with cesium iodide (for claims 24 -27;

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halide of an alkali metal for claims 12-23; see lines 37-46 of col.6, lines 16-20 of col.5 and claim 7 of Leblans) in order to stabilize the device in which the phosphor is employed against humidity (see lines 1-5 of col.3).

Therefore, it would have been obvious to one of ordinary skill in the art, at the time of the invention, to modify the luminescent layer, as disclosed by Leblans, in the electroluminescence device of Kido in order to stabilize the device in which the phosphor is employed against humidity.

Further regarding claims 12-13, wherein the inorganic compound is a metal halide, Leblan teaches that the inorganic compound for the luminescent material can be a metal halide comprising of a halide of a rare earth element with a halide of an alkali metal (see lines 37-46 of col.6, lines 16-20 of col.5 and claim 7 of Leblans which discloses combination of Europium II bromide and Cesium lodide).

Note: Applicant's specification discloses that "More specifically, examples of the "metal halide" include "combinations of a halide of a rare earth element with a halide of an alkali metal or alkaline earth metal". (For example, Applicant's spec. paragraph [0075]). Therefore, claims 12-15 are disclosed by Leblins that teaches that the inorganic compound is Europium II bromide and Cesium lodide; which have in turn been defined as metal halides in the Applicant's disclosure.

#### Other Prior Art Cited

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure:

**DE010242006 is** in the patent family of 7038221 teaches that a luminophore plate comprises of CSI and Europium Bromide.

US 5540859 teaches an electroluminescent material comprising of CSI and Europium Bromide.

US 4023039 teaches that the luminescent metal compound is Cesium lodide.

### Response to Arguments

The arguments filed on 10/20/08 have been considered.

The Applicant has made the following arguments:

- 1. The amended portion of the claim 1 recites the limitation of a "luminescent layer formed only of an inorganic compound" (see page 7 and pages 10-11 of the Remarks).
- 2. Receipt of Foreign priority documents by the Office (see page 9 of the Remarks).
- 3. Leblans is directed to "X-ray intensifying screens", in which phosphor particles absorb X-rays and convert them into visible light or ultraviolet radiation and therefore Leblans does not disclose a screen or a layer that undergoes luminescence by recombination of a hole and an electron, namely electroluminescence (see pages 11 and 12 of the Remarks)

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The following are the response to the Arguments:

1. The arguments are moot in view of new grounds of rejection using the new prior art

of Watanabe and Lee.

2. The argument is considered and it is hereby confirmed that the foreign priority

papers have been all received.

3. The arguments are not persuasive since the Applicant is directed to section 2112 [R-

3] I of MPEP, wherein it is stated that I. **SOMETHING WHICH IS OLD DOES NOT** 

BECOME PATENTABLE UPON THE DIS-COVERY OF A NEW PROPERTY "[T]he

discovery of a previously unappreciated property of a prior art composition, or of a

scientific explanation for the prior art's functioning, does not render the old composition

patentably new to the discoverer." Therefore a composition that has already been

disclosed by Leblans as a **phosphor** does not become patentable upon the

discovery of a new property of the same phosphor that undergoes a hole and electron

recombination.

Therefore the claims are not in condition for allowance.

Conclusion

1. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time

policy as set forth in 37 CFR 1.136(a).

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A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

#### **Contact Information**

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Fatima Farokhrooz whose telephone number is (571)-272-6043. The examiner can normally be reached on Monday- Friday, 9 am - 5 pm. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Minh-Toan Ton can be reached on (571) 272-2303. The fax phone number for the organization where this application or proceeding is assigned is (571) 273-8300. Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <a href="http://pair-direct.uspto.gov">http://pair-direct.uspto.gov</a>. Should you

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have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/Fatima N Farokhrooz/ Examiner, Art Unit 2889

/Toan Ton/ Supervisory Patent Examiner, Art Unit 2889